Robust Biocomposite Film of Polylactic Acid and Ferroferric Oxide as a Radar Absorbing Material

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Abstract: The polymer/ferroferric oxide (Fe₃O₄) foam and aerogel composites generally exhibit superior radar absorptivity performance. However, these composites have poor mechanical and thermal properties. This study manufactured a polylactic acid (PLA)/Fe₃O₄ bio-composite and evaluated the radar absorptivity, thermal, and mechanical properties of radar-absorbing material. The composites were prepared using a solvent casting method to mix PLA and Fe₃O₄ at varying concentrations, followed by evaporation, oven drying, and hot pressing into a film. Thermogravimetric analysis showed that the decomposition temperature of the PLA/Fe₃O₄-5% composite occurred at around 306 °C, which shifted to a lower decomposition temperature of PLA. The addition of 25 wt.% Fe₃O₄ improved the tensile modulus of neat PLA from 2.92 to 3.55 GPa. The vector network analyzer demonstrated that the addition of Fe₃O₄ at 25% improved the reflection loss of PLA from -5.17 to -25.83 dB at a thickness of 3 mm, with energy absorbed by 99.74% at frequency position 10.58 GHz. These results demonstrated that PLA/Fe₃O₄ composites have great potential in radar-absorbing practical applications.

Keywords: radar absorptivity; thermal properties; mechanical properties; reflection loss; *PLA/Fe₃O₄ biocomposites*

INTRODUCTION

Nowadays, radar absorbing materials (RAM) have been massively developed in military equipment stealth applications to minimize reflected electromagnetic waves to the radar system [1-2]. Therefore, RAM, with its high absorptivity, ultra-thin, superior tensile strength, and good thermal stability, has attracted considerable research interest [3-4]. The basic principle of microwave absorption is to convert microwave energy into heat energy through various absorption mechanisms. Microwave absorbers were generally manufactured from carbon-based materials, conductive polymer-based materials, ceramics, and magnetic materials [5]. Various composites, including solid films, foams, and aerogels, were designed to enhance the absorptivity performance of radar waves. Foam or aerogel is one of the lightest materials because it is filled with air in the pores or internal cavities, generally prepared through freeze drying [6].

Ferroferric oxide (Fe₃O₄) is a magnetic material with excellent permeability and complex permittivity, the parameters that govern the significance of microwave absorption. Other advantages of Fe₃O₄ are high saturation magnetization value, high curie temperature, facile synthesis, and low cost [7-8]. Recently, researches have been directed toward increasing the microwave absorption of pure Fe₃O₄ particles, mainly based on particle sizes, including micro-spheres [9], nano-spheres [10], and nanocrystals [11]. The Fe₃O₄ micro-sphere synthesized by a simple chemical at 90 °C showed a reflection loss value of -45.2 dB at 4.67 GHz; meanwhile, Fe₃O₄ nano-spheres and nanocrystals each showed a reflection loss (RL) of -21.2 dB at 5.6 and 8.16 GHz. However, Fe₃O₄ as an absorbent material has some limitations, including a rapid decrease in the value of the permeability at microwave frequencies, large density, ease of oxidation, and impedance mismatch, which hinders the effective performance of microwave absorbers [12].

The radar absorptivity performance of foam and aerogel composites manufactured from Fe₃O₄ dispersion in a polymer matrix has been evaluated in several previous studies. The foam and aerogel composites showed higher radar absorption values, low density, and good flexibility. Phadtare et al. [13] reported superior radar absorption performance of a Fe₃O₄/divinylbenzene/ethyl hexyl acrylate foam composite with an RL value of -23 dB at 9.3 GHz frequency position in 15 wt.% Fe₃O₄ filler concentration. In another study, Jiang et al. reported an RL value of -26.45 dB for the melamine/Fe₃O₄ foam composite at the frequency of 7.76 GHz with a filler ratio of 30 wt.% [14]. The large internal cavity will cause multireflection and polarization loss; therefore, more radar energy is trapped in the composite. The large internal cavity provides a larger surface area, which facilitates impendence matching and polarization loss through multi-reflections, causing attenuation of the radar wave in the material. These foam and aerogel composites have limitations in a broader of applications.

Polylactic acid (PLA) is a bio-based polymer with excellent mechanical and thermal properties compared with synthetic polymers such as polyethylene. Notably, PLA is one of the most commonly used biopolymers for various applications because of its ease of processing [15]. PLA has been prepared with various types of inorganic fillers for numerous applications, one of which is to improve thermal and mechanical properties [16]. PLA is a biodegradable plastic that has great potential to replace petroleum-based fuel plastics due to its high stiffness and strength [17].

The objective of this study is to evaluate the radar absorptivity, thermal and mechanical properties of

PLA/Fe₃O₄ composites. The composites were prepared by mixing PLA and 5 to 25 wt.% Fe₃O₄ by solvent casting. The radar absorption performance of PLA/Fe₃O₄ composites was evaluated using a Vector Network Analyzer (VNA). Meanwhile, the thermal and mechanical properties of the composites were determined using thermogravimetric analysis and a universal testing machine, respectively. In addition, the morphological, crystal structure, and magnetic properties of the film composite were also investigated using Scanning Electron Microscope (SEM), X-ray Diffraction (XRD), and vibrating sample magnetometer (VSM).

EXPERIMENTAL SECTION

Materials

PLA 4043D was supplied by Natureworks Co., Minnetonka, MN (USA). Dichloromethane (DCM, \geq 99.0%), ammonia solution 25%, and iron (III) chloride hexahydrate (FeCl₃·6H₂O) were purchased from Merck KGaA, Germany. Ferrous sulfate heptahydrate (FeSO₄·7H₂O, \geq 99.0%) was supplied by HiMedia Laboratories Pvt.Ltd., India.

Instrumentation

The surface morphology of Fe_3O_4 , neat PLA, and PLA/Fe₃O₄ composites was observed under SEM (JEOL JSM-IT200, Japan) operating at 3.0 kV. The sample of Fe₃O₄ was gold-coated to improve conductivity before observation. The crystal structure of the samples was evaluated by XRD (Shimadzu 7000, Japan) with an X-ray tube of Cu operating at 30.0 mA and 40.0 kV. The XRD pattern of samples was recorded in the interval of 10° to 80° at a speed of 2°/min. The degree of crystallinity (X_c) is calculated using the formula [19]:

$$Xc = \frac{\text{area of crystalline peaks}}{\text{area of all peaks (crystalline + amorphous)}} \times 100\%$$

The thermal stability of neat PLA and PLA/Fe₃O₄ composites was evaluated using PerkinElmer TGA 4000, USA. The samples were immersed in a closed aluminum pan. Temperature programs were from 25 to 600 °C at a temperature increase speed of 10 °C/min under a nitrogen condition. The heating scan of the specimens was used to estimate thermal parameters such as

decomposition temperature (T_d), temperature for 50% weight loss (T₅₀), and residual weight (remaining Fe₃O₄ content) at 600 °C. A universal testing machine (Shimadzu AGS-X series 10kN, Japan) was used to investigate the mechanical properties of the samples. The tensile testing samples were fabricated using a mini-jet pro to obtain a dumbbell-shaped tensile bar (ASTM D638-14 type V). The sample gauge length, width, and thickness were approximately 30, 3, and 3 mm, respectively, measured by a caliper when gripped. The crosshead rate was set at 1 mm/min. All data on mechanical properties were recorded in the 5 measurements. The magnetic hysteresis loops were investigated by a vibrating sample magnetometer (VSM250 Dexing Magnet Ltd) with an applied magnetic field of 1 tesla under ambient temperature. The sample dimensions prepared were $5 \times 5 \times 1$ mm (length-width-thick). The radar absorption of the samples was analyzed using the waveguide method at a frequency of 8.0 to 12.0 GHz using a vector network analyzer (VNA) Anritsu MS46322A, Japan, at room temperature. which The sample, measured $30 \times 30 \times 3$ mm (length, width, thickness) was placed on the port and coated with a metal reflector. The complex permeability and air transmittance were measured to calibrate the instrument, which found results of μ " = ϵ " = 1 and $\mu' = \epsilon' = 0$. The reflection loss was calculated using line transmitting theory by adhering to the processes outlined in prior publications [20].

Procedure

The preparation of Fe₃O₄ particles was carried out following the procedure of the previous study [18]. A slight modification was made by extending the stirring time of 5.406 g (0.02 mol) FeCl₃·6H₂O and 2.701 g (0.01 mol) FeSO₄·7H₂O from 30 min to 2 h. PLA (described in Table 1) was dissolved in a beaker using DCM under stirring for 4 h. The Fe₃O₄ particles were added gradually to the PLA solution after the PLA was completely dissolved, and the stirring was kept going for another 4 h. The PLA/Fe₃O₄ mixture was poured and dispersed on trays, and the dichloromethane solvent was evaporated at ambient temperature for 12 h before oven drying for 8 h at 80 °C. The dried mixture was cut into small

Table 1. The composition of the samples				
Sampla nama	Fe_3O_4	PLA	DCM	
Sample name	(g)	(g)	(mL)	
Neat PLA	0	10	100	
PLA/Fe ₃ O ₄ -5%	0.5	9.5	100	
PLA/Fe ₃ O ₄ -10%	1.0	9.0	100	
PLA/Fe ₃ O ₄ -15%	1.5	8.5	100	
PLA/Fe ₃ O ₄ -20%	2.0	8.0	100	
PLA/Fe ₃ O ₄ -25%	2.5	7.5	100	

pieces of around 5×5 mm and hot-pressed at a temperature of 155 °C with a pressure of 50 MPa for 10 min to obtain a PLA/Fe₃O₄ composite film of 3×3 cm with various thickness.

RESULTS AND DISCUSSION

Morphological Analysis

The morphology of Fe₃O₄ is shown in Fig. 1(a). The SEM image shows Fe₃O₄ is a microparticle with an irregular shape with an average size of fewer than 5 μ m. The previous co-precipitation method produced similar findings [18]. In the Fe₃O₄ particle preparation process, several factors influence particle size and particle distribution: stirring speed, ambient temperature, and solution, the concentration of ammonia solution, and the Fe²⁺ and Fe³⁺ reagents used [21].

The fracture morphology of neat PLA and PLA/Fe₃O₄ composites observed using SEM is shown in Fig. 1(b-g). It can be seen from the figure that the pure PLA fracture surface is caused by the insoluble PLA portion and the appearance of pores due to DCM evaporation. The addition of more fillers results in the formation of agglomerates. Fe₃O₄ particles were visible in the fine-grained composite, which spread uniformly in the PLA matrix with a diameter smaller than $5\,\mu m$ even in the Fe₃O₄ concentration of 25 wt.%. At 5 wt.% Fe₃O₄ concentration, there were no obvious cracks or holes on the fracture surface of the composite, which means that good interfacial interactions occur between the PLA matrix and Fe₃O₄ particles. With the addition of Fe₃O₄ concentration in the composite, the holes or cracks are more clearly visible, which means that the interfacial interaction is getting weaker.

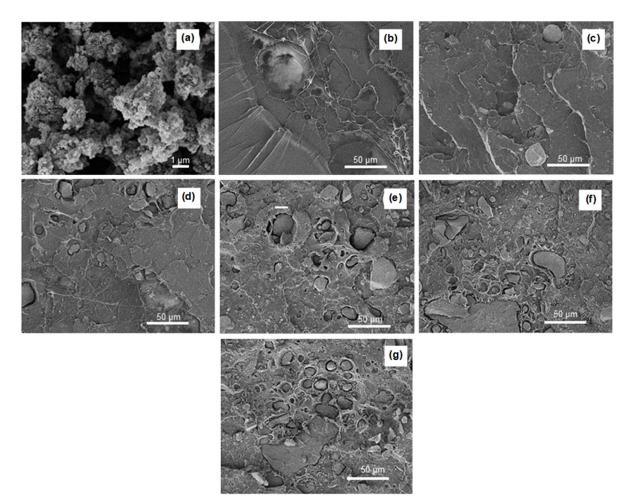


Fig 1. SEM images of morphology: (a) Fe_3O_4 particles; fracture of all composites: (b) Neat PLA, (c) PLA/Fe₃O₄-5%, (d) PLA/Fe₃O₄-10%, (e) PLA/Fe₃O₄-15%, (f) PLA/Fe₃O₄-20%, and (g) PLA/Fe₃O₄-25%

XRD Analysis

The diffraction peaks in the XRD pattern of Fe_3O_4 , neat PLA, and PLA/Fe₃O₄ composites are shown in Fig. 2. The six diffraction peaks of Fe₃O₄ crystal at 2 θ values are 30.3742, 35.6851, 43.3585, 53.8291, 57.5713, and 62.9956, according to the values of the index Miller of (220), (311), (400), (422), (511) and (440) crystal planes from the inverted cubic spinal structure of Fe₃O₄ (JCPDS 65-3107) [22-23]. These peaks indicated that the prepared Fe₃O₄ formed spinal structures, and no characteristic impurity peaks were detected in the XRD pattern.

Fig. 2 shows neat PLA exhibits large diffraction peaks at 2θ values of 16.52 and 32, which correspond to (200/110) and (203) according to a previous study [24]. When Fe₃O₄ particles were added, all composites displayed

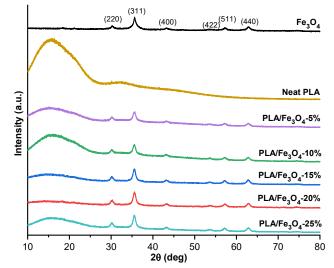


Fig 2. XRD patterns of Fe₃O₄, neat PLA, and PLA/Fe₃O₄ composites

broad diffraction peaks similar to the neat PLA peaks. The intensity of the PLA diffraction peaks did not increase with adding more Fe_3O_4 in the composite, although those shrunk and widened. The Fe_3O_4 addition may have prevented the development of numerous crystals and the organization of PLA molecular sequences, but it did not affect the crystal structure of PLA. Based on the calculation, X_c of neat PLA, PLA/Fe₃O₄-5%, PLA/Fe₃O₄-10%, PLA/Fe₃O₄-15%, PLA/Fe₃O₄-25%, respectively, is 28.90, 32.69, 33.27, 34.48, 35.62, and 37.56%.

The Thermal Stability of PLA/Fe₃O₄ Composites

The TGA test was used to investigate the effect of Fe₃O₄ on the thermal stability of PLA. Fig. 3 shows the weight loss of PLA and PLA/Fe₃O₄ composites at varying Fe_3O_4 content as a function of temperature. The T_d , T_{50} , and residual Fe₃O₄ content at 600 °C are listed in Table 2. Based on the thermogram, we can see that neat PLA has thermal stability up to a degradation onset temperature of about 320 °C. Then, the decomposition temperature occurs at around 349 °C. After reaching it, the sample weight decreases drastically by a single-stage decomposition until the maximum rated temperature of 380 °C. The weight loss of the PLA continues to decrease slowly until the final test temperature, where the residual weight is 0.178%. The interactions between volatile organic compounds are primarily responsible for keeping the char structure firm. This thermal behavior is typical for polylactic acid polymers [25-26].

The presence of 10 wt.% Fe_3O_4 on the PLA could maintain thermal stability up to a temperature of 280 °C. The decomposition temperature of the PLA/Fe₃O₄-10% composite occurred at around 293 °C, which shifted to a lower decomposition temperature of PLA. Interestingly, the addition of a higher Fe₃O₄ concentration did not significantly change the decomposition temperature of the PLA composite. After the decomposition temperature, the weight of all composites dropped drastically until around 30 to 50% of the sample weight remained at 300 °C. The weight loss of the PLA composites continued decreasing slowly until the final test temperature. The visible changes were in the remaining weight of each composite, where the final weight was determined based on the concentration of Fe₃O₄. Considering the physicochemical properties and the chemical structure of the Fe₃O₄, it can be assumed that the observed reduction in thermal stability is related to two simultaneous mechanisms. A similar effect is observed in the case of thermal stability of the other PLA/inorganic-filler composites, where the presence of copper slag contributes to a reduction in the decomposition temperature of PLA [25].

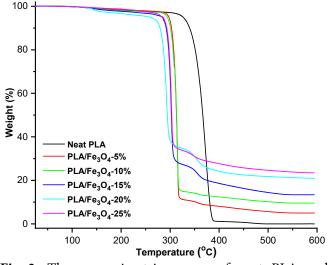


Fig 3. Thermogravimetric curves of neat PLA and PLA/Fe₃O₄ composites with different Fe_3O_4 contents (wt.%)

Samples	T _d (°C)	T _{0.50} (°C)	Residual weight (%)
Neat PLA	349.16	365.13	0.178
PLA/Fe ₃ O ₄ -5%	306.14	312.92	5.020
PLA/Fe ₃ O ₄ -10%	304.23	312.36	9.707
PLA/Fe ₃ O ₄ -15%	297.14	303.32	13.390
PLA/Fe ₃ O ₄ -20%	258.23	288.22	20.896
PLA/Fe ₃ O ₄ -25%	295.85	302.41	23.355

Table 2. TGA results of neat PLA and PLA/Fe₃O₄ composites

The Mechanical Properties of PLA/Fe₃O₄ Composites

A tensile test was used to evaluate the influence of Fe_3O_4 addition on the mechanical characteristics of PLA. Table 3 and Fig. 4 present the mean and standard deviation of the mechanical properties of all the samples. The table and figure clearly show the tensile modulus of PLA improved with an increase of Fe_3O_4 content while the tensile strength and strain at break decreased. The addition of 5 wt.% Fe_3O_4 improved the modulus from 2.92 to 3.09 GPa. Furthermore, the composite stiffness increased significantly with the addition of a higher Fe_3O_4 concentration. The composite reached the highest stiffness of 3.55 GPa at a 25 wt.% Fe_3O_4 content.

It is clearly shown that added Fe₃O₄ deteriorated the composite strength compared to neat PLA. The decrease in tensile strength of PLA occurred after the addition of 5 wt.% Fe₃O₄ content from 67.57 to 57.94 MPa. Therefore, we could see that the strength remains almost constant between 5 to 25 wt.% Fe₃O₄ content. This decrease in the tensile strength of the composites was caused by poor interface adhesion between Fe₃O₄ particles and PLA. The weakened ion-dipole interaction between the ferrite cation and oxygen atom in the PLA carbonyl group also affects to decrease of tensile strength. The strain at the break value of neat PLA increased almost two times with the addition of 5% by weight of Fe₃O₄ from 3.52 to 5.21%. Then it decreased with the addition of Fe₃O₄ filler and reached 2.08% for 5 wt.% filler content. A decrease in elongation at break with the addition of Fe₃O₄ to polymers is commonly observed in thermoplastic composites. In brief, the mechanical properties of these composites depend on the following factors: the adhesion between the PLA matrix and the Fe₃O₄ filler, the ratio of the Fe₃O₄ filler to the PLA polymer, the size of the Fe₃O₄ filler, the type and characteristics of the filler, and the degree of crystallinity of the matrix [16]. Decreased interphase stress results in reduced bond strength of the polymer structure. In addition, low adhesion between the matrix and filler results in cavities at the filler matrix interface resulting in a change in the strength value [27]. The tensile modulus value describes the stiffness of a material that allows it to withstand loads without shape deformation. High tensile strength and modulus values are needed for large loads [28].

Magnetic Property Analysis of PLA/Fe₃O₄ Composites

The hysteresis loops of neat PLA and PLA/Fe₃O₄ composite are shown in Fig. 5(a). The magnetic properties such as saturation magnetization (M_s), remanence magnetization (M_r), and coercive field (H_c) are listed in Table 4. A magnet attraction test was performed to determine the magnetic response of neat PLA and PLA/Fe₃O₄ composite. Neat PLA has no response,

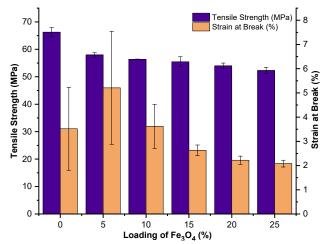


Fig 4. The graphs of tensile strength and strain at break depend on variations in Fe_3O_4 loading

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Samples	Tensile modulus	Tensile strength	Strain at break	
Samples	(GPa)	(MPa)	(%)	
Neat PLA	2.92 ± 0.13	66.27 ± 1.70	3.52 ± 1.71	
PLA/Fe ₃ O ₄ -5%	3.09 ± 0.05	57.94 ± 0.81	5.21 ± 2.33	
PLA/Fe ₃ O ₄ -10%	3.16 ± 0.06	56.37 ± 0.08	3.62 ± 0.91	
PLA/Fe ₃ O ₄ -15%	3.41 ± 0.06	55.48 ± 1.82	2.63 ± 0.22	
PLA/Fe ₃ O ₄ -20%	3.53 ± 0.14	53.97 ± 1.04	2.22 ± 0.17	
PLA/Fe ₃ O ₄ -25%	3.55 ± 0.07	52.28 ± 1.16	2.08 ± 0.14	

Table 3. Mechanical properties of neat PLA and PLA/Fe₃O₄ composites

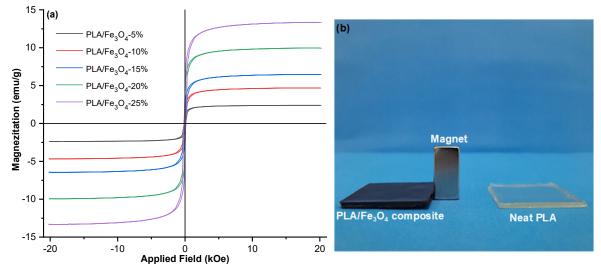


Fig 5. (a) Magnetic properties of PLA/Fe₃O₄ composites with a Fe₃O₄ content, (b) Comparison of magnet attraction by neat PLA and PLA/Fe₃O₄ composite

Table 4. Magnetic parameter of PLA/Fe₃O₄ composites

Samples	M _s (emu/g)	M _r (emu/g)	H _c (kOe)
Neat PLA	-	-	-
PLA/Fe ₃ O ₄ -5%	2.39	0.16	20.199
PLA/Fe ₃ O ₄ -10%	4.68	0.31	20.161
PLA/Fe ₃ O ₄ -15%	6.46	0.41	20.215
PLA/Fe ₃ O ₄ -20%	9.94	0.63	20.223
PLA/Fe ₃ O ₄ -25%	13.31	0.84	20.250

while PLA/Fe_3O_4 composite provides a magnetic response, as shown in Fig. 5(b).

The addition of 5 wt.% Fe₃O₄ into PLA resulted in magnetic properties of the PLA composite with a M_s of 2.39 emu/g and a M_r value of 0.16 emu/g (Table 4). The M_s and M_r values of the composite increased proportionally with the increase in Fe₃O₄ concentration. At 25 wt.% Fe₃O₄ content, the maximum M_s and M_r obtained were 13.31 and 0.84 emu/g, respectively. All composites' coercivity did not change significantly and remained almost constant with increasing Fe₃O₄ content. It shows that the coercive behavior is not controlled by the content of Fe₃O₄ particles but instead by the type of magnetic particles. The trend of increasing M_s and M_r due to the addition of magnetic filler is following previous studies [29].

Radar Absorption Performance of PLA/Fe₃O₄ Composite

The radar absorption performance of the samples is

described by the RL value. Fig. 6 shows the RL curve of neat PLA and PLA/Fe₃O₄ composites, while the reflectivity, energy absorbed, and the position of the maximum absorption frequency is presented in Table 5. Although neat PLA resulted in an RL value of 5.17 dB, PLA is not categorized as a radar-absorbing material. The radar absorption value that appears from the neat PLA is perhaps due to the component of carbon atoms in PLA, trapped air, and sample thickness. However, when PLA was combined with Fe₃O₄ at the same thickness, the radar absorption behavior increased significantly as the Fe₃O₄ concentration increased. The addition of 5 wt% Fe₃O₄ improved the RL of PLA from -5.17 to -15.75 dB, with energy absorbed of 97.32% at frequency position 10.78 GHz. Furthermore, the RL of composites increased proportionally with the increase of Fe₃O₄ concentration. The composite at 25 wt.% Fe₃O₄ reached the highest RL of -25.83 dB, with energy absorbed 99.74% at 10.58 GHz and thickness of 3 mm.

As a magnetic loss absorber, the saturation magnetization of material will affect the radar absorption performance. The radar absorption intensity will grow as the saturation magnetization value increases. The addition of Fe_3O_4 magnetic material to PLA enhances the saturation magnetization of the composite, increasing the absorption performance of radar signals, as demonstrated in Tables 5 and 6. A comparison of the

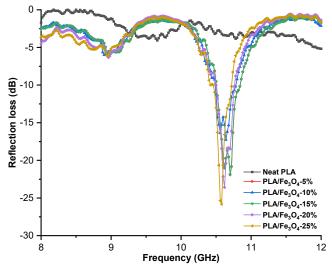


Fig 6. Radar absorption properties (thickness = 3 mm) of neat PLA and PLA/Fe₃O₄ composites. In set shows the zoom-out of the reflection loss graph at 10 to 11 GHz

reflectivity of composites of Fe_3O_4 in various polymer matrices is presented in Table 6. The RL value of the Fe_3O_4 in the PLA matrix is superior to other Fe_3O_4 composites. With a load of only 25 wt.% Fe_3O_4 , PLA composite can absorb radar 72% higher than Poly(3,2ethylenedioxythiophene) (PEDOT) composite at 80 wt.% Fe_3O_4 content. It can be concluded that the magnetic properties of Fe_3O_4 remained superior in the PLA polymer.

CONCLUSION

The radar absorption, thermal, and mechanical properties of PLA/Fe₃O₄ biocomposite have been successfully evaluated. The vector network analyzer measurements showed that the highest reflection loss of -25.83 dB occurred in PLA composites with a Fe₃O₄ concentration of 25 wt.%, about 399% higher than neat PLA. The TGA measurement revealed that Fe₃O₄ in various concentrations was able to maintain the thermal stability of neat PLA up to 280 °C. However, the decomposition temperature of all composites shifted to a lower temperature of neat PLA. The addition of 25 wt.% of Fe₃O₄ in PLA increased the tensile modulus of neat PLA from 2.92 to 3.55 GPa, while the tensile strength and the strain at the break did not change significantly. Based on these results, PLA/Fe₃O₄ biocomposites offer great promise in a wide range of radar absorption applications due to their good RL and thermo-mechanical properties.

Table 5. The reflectivity, energy absorbed, and the position of the maximum absorption frequency of neat PLA and PLA/Fe_3O_4 composites

Samples	Reflectivity (dB)	Energy absorbed (%)	Frequency position (GHz)
Neat PLA	-5.17	69.59	12.00
PLA/Fe ₃ O ₄ -5%	-15.72	97.32	10.66
PLA/Fe ₃ O ₄ -10%	-17.27	98.03	10.64
PLA/Fe ₃ O ₄ -15%	-21.90	99.35	10.70
PLA/Fe ₃ O ₄ -20%	-23.58	99.56	10.62
PLA/Fe ₃ O ₄ -25%	-25.83	99.74	10.58

Polypyrrole 50 -22.4 12.9 2.3 [31] divinylbenzene (DVB), and ethyl hexyl acrylate (EHA) 15 -23 9.3 3 [13] Polyaniline 62.5 -24.7 14.8 2.0 [32] Polyethylene 10 -24.3 9 4 [33]		1	1	1		
(wt.%) absorption (dB) position (GHz) (mm) Poly(3,2-ethylenedioxythiophene (PEDOT) 80 -15 9 2.8 [30 Polypyrrole 50 -22.4 12.9 2.3 [31 divinylbenzene (DVB), and ethyl hexyl acrylate (EHA) 15 -23 9.3 3 [13 Polyaniline 62.5 -24.7 14.8 2.0 [32 Polyethylene 10 -24.3 9 4 [33	Dolymor motrix	Fe ₃ O ₄ loading	Maximum	Frequency	Thickness	Def
Polypyrrole 50 -22.4 12.9 2.3 [31] divinylbenzene (DVB), and ethyl hexyl acrylate (EHA) 15 -23 9.3 3 [13] Polyaniline 62.5 -24.7 14.8 2.0 [32] Polyethylene 10 -24.3 9 4 [33]		(wt.%)	absorption (dB)	position (GHz)	(mm)	Kel.
divinylbenzene (DVB), and ethyl hexyl acrylate (EHA) 15 -23 9.3 3 [13] Polyaniline 62.5 -24.7 14.8 2.0 [32] Polyethylene 10 -24.3 9 4 [33]	Poly(3,2-ethylenedioxythiophene (PEDOT)	80	-15	9	2.8	[30]
Polyaniline62.5-24.714.82.0[32Polyethylene10-24.394[33	Polypyrrole	50	-22.4	12.9	2.3	[31]
Polyethylene 10 -24.3 9 4 [33	divinylbenzene (DVB), and ethyl hexyl acrylate (EHA)	15	-23	9.3	3	[13]
	Polyaniline	62.5	-24.7	14.8	2.0	[32]
PLA/Fe ₃ O ₄ -25% 25 -25.83 10.58 3 This wa	Polyethylene	10	-24.3	9	4	[33]
	PLA/Fe ₃ O ₄ -25%	25	-25.83	10.58	3	This work

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AUTHOR CONTRIBUTIONS

Rafles Sinaga conducted the conceptualization, data curation, formal analysis, investigation, methodology, and writing-original draft. Wida Banar Kusumaningrum conducted the conceptualization, investigation, project administration and resources. Yana Taryana conducted data curation, formal analysis, investigation, and methodology. Widya Fatriasari conducted a review, and formal analysis. Zuratul Ain Abdul Hami conducted a review, methodology, formal analysis. Holilah conducted formal analysis, investigation, methodology, and writingoriginal draft. Yudi Darma conducted conceptualization, supervision, and writing-original draft. Lisman Survanegara conducted the conceptualization, data curation, formal analysis, funding acquisition, investigation; methodology, project administration, resources, supervision, and writing-original draft.

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