CHARACTERISTICS OF HYDROTHERMAL ALTERATION IN CIJULANG AREA, WEST JAVA, INDONESIA

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Abstract

Characterization of hydrothermal alteration in the Cijulang area (West Java, Indonesia) was carried out using shortwave infrared spectroscopy. Hydrothermal alteration in the Cijulang area occurs in the calc-alkaline volcanic and volcaniclastic rocks. Shortwave infrared spectroscopic measurements of reflectance for altered rocks and minerals were carried out by ASD-FieldSpec and the laboratory spectra acquired were then analysed with “The Spectral Geologist” software program. Shortwave infrared spectroscopy is capable of detecting most fine-grained alteration minerals from different hydrothermal alteration zones. Characteristic alteration minerals identified from the SWIR technique include pyrophyllite, alunite, kaolinite, dickite, illite, montmorillonite, polygorskite, gypsum, epidote, paragonite, and muscovite. Most of the spectra show mixture of alteration minerals and only a few display pure spectra of single mineral. The crystallinity of kaolinite from the samples was also determined from the reflectance spectra and show moderately to high crystallinity. Alteration system of the Cijulang prospect is similar to others documented high-sulfidation epithermal deposits, such as Rodalquilar (Spain), Summitville (Colorado), and Lepanto (Philippines). A characteristic alteration sequence and zonation of advanced argillic, argillic and propylitic alteration outward from the silica core has resulted from the progressive cooling and neutralization of hot acidic magmatic fluid with the host rocks.

Keywords: Cijulang, High-sulfidation, Alteration minerals, Shortwave Infrared Spectroscopy

1 Introduction

Shortwave infrared reflectance (SWIR) spectroscopy has been widely used in hydrothermal alteration mapping of various types of mineral deposits, such as high- and low-sulfidation epithermal, mesothermal, porphyry, sediment-hosted gold and copper, uranium, volcanogenic massive sulfide (VMS) and kimberlite. SWIR spectroscopy involves detection of the energy generated by vibrations within molecular bonds in the 1300- to 2500-nm (SWIR) range of electromagnetic spectrum (Thompson et al., 1999). Most minerals exhibit characteristic spectrum and major diagnostic absorption features in this SWIR region as a result of bending and stretching of certain molecules and radical, including OH, H2O, NH4, and Cation-OH bonds (such as AlOH, MgOH and FeOH) (Pontual et al., 1997a; Thompson et al., 1999). The spectral absorption features associated with different silicates occur at or near 1400nm (OH and water) and 1900nm (water) whereas other important

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and diagnostic spectral absorption features related to cation-OH bonds such as AlOH, FeOH and MgOH occur at or near 2200nm, 2250nm and 2330nm respectively (Figure 1). The absorption features that represent these bonds or mineral groups are characteristic of hydrothermal alteration and mineral groups belonging to kaolinite, halloysite, pyrophyllite, smectite clays, dickite, micas, chlorites, alunite, jarosite, calcite (Pontual et al., 1997).

Reflectance spectra can be acquired by field portable spectrometers, such as GER-IRIS, ASD-FieldSpec and PIMA or by airborne remote sensing. Application of spectrometers allows rapid identification of the fine-grained hydrothermal alteration minerals which are generally difficult to identify by conventional macroscopic observation. SWIR spectroscopy not only detects the presence of minerals in the sample but also determines the relative abundance of mineral within the samples. Moreover, mineralogical and compositional variation in certain specific mineral species can also be recognized by the SWIR technique (Thompson et al., 1999).

Cijulang area is located in the Garut Regency of West Java, Indonesia (Figure 2) and the prospect has been explored by PT Aneka Tambang since 1994. This research aims to explore the characteristics of the hydrothermal alteration associated with pyrite-enargite-gold mineralization in the Cijulang prospect by the application of reflectance spectroscopy.

2 Research methods

Field investigation was carried out in the Cijulang area of West Java, Indonesia during March, 2013. A total of 70 altered rock and clay samples were collected from different hydrothermal alteration zones. Study on alteration mineralogy was carried out by Shortwave Infrared Spectroscopy (SWIR) aided by petrographic microscopy, X-Ray Diffraction and SEM-EDS analyses.

Shortwave infrared (SWIR) spectroscopic measurement of reflectance for altered rocks and clay minerals was carried out by Analytical Spectral Devices (ASD-FieldSpec), a spectro-radiometer which collected the reflectance in the 350 to 2500nm spectral range. The reflectance spectra acquired by ASD device were then analyzed with “The Spectral Geologist” (TSG)” software. The TSG provides automated assistance in the mineral identification and statistical analysis of the spectra. Mineral identification was conducted by spectral matching with reference library spectra. The Spectral Assistant (TSA) in The Spectral Geologist uses a library of 551 spectra of pure minerals and water. The software identifies a maximum of three minerals assigning relative weights and an error of matching. The relative weight does not represent the abundance of minerals in the samples but it reflects the relative proportion of one mineral to another which is influenced by grain size and distribution of the minerals. The level of confidence in the mineral identification is indicated by the error of matching.

Interpretation of spectral data from TSG software were cross-checked by spectral interpretation field manual and USGS library spectra. Identification of the mineral is based on the following spectral characteristics; wavelength position, intensity and shape of the absorption troughs and the overall shape of the entire spectrum. Shifting of the wavelength positions of diagnostic absorption feature represents compositional variation in minerals whereas variation in depth or width of the absorption feature reflects the variation in crystallinity or the grain size of the relative abundance of mineral (Pontual et al., 1997).

Bulk rock powdered and clay fraction samples were analyzed by Rigaku RINT-2100 Diffractometer at Laboratory of Earth Resources Engineering Department, Faculty of Engineering, Kyushu University. X-ray diffraction analysis was done using CuKα radiation at 40kV and 20mA. Microprobe examination of alteration minerals was carried out by SHIMADZU SS-550 Scanning Electron Microscope equipped with a genesis-2000 EDX Spectrometer at the Centre for Advanced Instrumental Analysis, Kyushu University, Japan. Experimental condition for the analysis was performed at an accelerating potential of 15kV, beam current about 10mA, and 3μm beam diameter.
Figure 1: Major spectral absorption features in SWIR range (1300-2500nm) (Pontual et al., 1997a).
3 Results and Discussion

3.1 Alteration Minerals

Hydrothermal alteration minerals identified from reflectance spectra of altered rocks and clay minerals include kaolinite, dickite, pyrophyllite, illite, muscovite, chlorite, alunite, epidote, polygorskite, and goethite. Most of the spectra show a mixture of these alteration minerals and only a few show pure spectra of single minerals. The common mineral assemblages identified from the reflectance spectroscopy include pyrophyllite–kaolinite, dickite–pyrophyllite, alunite–dickite, alunite–kaolinite, kaolinite–illite, illite–montmorillonite and chlorite–montmorillonite (Figures 3a and 3b).

3.2 SWIR Spectral Characteristics of Alteration Minerals

Kandite group minerals

Kaolinite \([\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4]\). Kaolinite generally occurs as pure mineral spectra or mix spectra with other mineral such as dickite, pyrophyllite and illite. Kaolinite spectra have the following major absorption features: hydroxyl \((\text{OH})\) stretching doublet around 1400nm and 1411nm, water \((\text{H}_2\text{O})\) absorption feature at 1913nm and Al-\text{OH} diagnostic double absorption feature at 2168nm and 2208nm (Figure 3a).

Dickite \([\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4]\). Dickite is one of the kandite group minerals commonly found in the Cijulang prospect. It occurs as pure mineral spectra or mix spectra with other minerals, such as pyrophyllite and kaolinite (Figure 3a). Spectral characteristics of dickite are similar to those of kaolinite. It has hydroxyl doublet absorption features occurring around 1390nm and 1415nm. Al-\text{OH} diagnostic double absorption features exhibiting at 2172nm and 2208nm, which are common characteristic of kandite group minerals (Figure 3a).

Pyrophyllite \([\text{Al}_2\text{Si}_4\text{O}_{10}(\text{OH})_2]\). Pyrophyllite spectra show absorption features at 1391–1396nm and 1411–1414nm due to vibration of...
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Figure 3: Reflectance spectra of alteration mineral mixtures from different hydrothermal alteration zones (a) advanced argillic and (b) argillic and propylitic. Kln-kaolinite, Prl-pyrophyllite, Dck-dickite, Alu-alunite, Ill-illite, Mnt-montmorillonite, Chl-chlorite, Ms-muscovite.

hydroxyl ions (OH\(^-\)) and water molecules (H-O-H), a doublet absorption feature at 2168nm and 2208nm due to vibrational processes associated with Al-OH bonds. Pyrophyllite is either seen as pure spectra or mixture with kaolinite (well crystalline) and dickite in the advanced argillically-altered samples (Figure 3a).

Smectite Group Minerals

Montmorillonite \([Na, Ca]_{0.33}(Al, Mg)_{2}(Si_4O_{10})(OH)_{2} \cdot nH_2O\]. Montmorillonite spectra have characteristic sharp minima and asymmetric shape of water absorption features at 1409–1411nm and 1911–1914nm (Figure 3b). A broad diagnostic Al-OH absorption feature generally occurs at 2208nm, a typical feature of aluminum-bearing clays (Hunt, 1979).

APS Group Minerals

Alunite \([K, Na]Al_3(SO_4)_2(OH)_6\]. Alunite is a diagnostic mineral of high-sulfidation epithermal system. Compositional variation in alunite-group minerals are recognized by shifts in the 1480nm spectral position, with value ranging from \(~1461\)nm (NH\(_4\)), to \(~1478\)nm (pure K), to \(~1496\) nm (Na) to 1510nm (Ca) (Thompson et al., 1999). Alunite from the Cijulang prospect is only found in the drill core section and generally associated with pyrite and quartz. Alunite is identified as potassium end-member (K-alunite) and its spectra have major characteristic alunite absorption features: hydroxyl (OH) stretching doublet occurring at 1428nm and 1480nm, diagnostic absorption feature at 1766nm and the Al-OH diagnostic absorption features at 2171nm, 2208µm and 2324nm (Figure 3a). Alunite generally occurs as mix spectra with other minerals, such as dickite and kaolinite.

Illite Group Minerals

Muscovite \([KAl_3(AlSi_3O_{10})(F,OH)_2]\]. White micas generally have absorption features occurring in the range between 2180nm to 2228nm due to the Al-OH bonds (Figure 3b). Varia-
tion in chemical composition strongly affects the position of absorption in white micas; Na-rich micas have absorption features around 2200nm, whereas Mg-FeOH-rich and potassic micas have similar features occurring in the ranges between 2216–2228nm and 2200–2208nm respectively (Pontual et al. 1997). Muscovite generally occurs as mixture with well crystalline kaolinite has the following spectral characteristic; hydroxyl (OH) absorption features at 1410nm, water (H$_2$O) absorption feature at 1917nm and Al-OH absorption feature at 2202nm. It is classified as Na-rich type by its Al-OH value occurring at 2200nm (Figure 3b).

**Illite** \[[K,H$_3$O](Al,Mg,Fe)$_2$(Si,Al)$_4$O$_10$\textit{(OH)}$_2$\textit{(H$_2$O)}\]. Illite has similar spectral characteristic to muscovite. Illite can be distinguished from muscovite by its deep absorption feature occurring at 1900µm due to molecular water within its crystal lattice (Figure 3b) (Pontual et al. 1997). Absorption feature occurring at ~2200nm reflects compositional variation in illite, those near 2190nm are more sodic whereas these feature close to 2206nm are more potassic (Halley, 2010).Illite from the Cijulang area is identified as paragonitic illite (Figure 3b).

**Other Minerals**

Other minerals such as Fe-chlorite, epidote, gypsum, palygorskite and goethite occur as mixture with other minerals in the analyzed samples. No pure spectra were obtained for these minerals and it was difficult to characterize their spectral characteristics. Anhydrous silicates and sulfides such as quartz and pyrite are difficult to identify because they show no absorption features in the shortwave infrared wavelength region. In addition, minerals present in amount less than 5% in the samples were not identified by the SWIR spectroscopy. The presence of such minerals was confirmed by the X-ray diffraction analysis.

**Kaolinite Crystallinity**

Kaolinite is the most common and abundant mineral in the hydrothermal alteration of the Cijulang area. So, it is important to determine the crystallinity of kaolinite. Kaolinite crystallinity determination was based on the analysis of shape and wavelength spacing of the Al-OH band doublets absorption features (Figure 4b) by using slope parameters in the 2160–2180nm wavelength region which is generally referred to as slope ‘2160’ and slope ‘2180’ parameters. Evaluation shows that the kaolinite from the Cijulang is characterized by moderate to high crystallinity (Figure 4b).

**3.3 Hydrothermal Alteration**

Hydrothermal alteration in the Cijulang area covers about 1×2 km$^2$ area of N–E trending elongate zone and hosted by andesitic lava, lapilli tuff and hydrothermal breccia. The distribution of alteration mineral assemblage indicates that Cijulang alteration is classified as epithermal high-sulfidation styles in which the silicic core is surrounded outward alteration mineral assemblages of advanced argillic, argillic and propylitic (Figure 5).

Silica alteration develops as both vuggy and massive silica. This alteration is easily recognized in the outcrop by weather-resistant bodies. Vuggy zone comprises residual silica remaining after loss of its reactive components by intense acid leaching. Massive silica is formed by secondary deposition of silica and later recrystallize to form quartz.

Advanced argillic alteration is the most widespread of alteration and identified by the presence of alunite, kaolinite, dickite, pyrophyllite, illite and pyrite in variable amount. The common mineral assemblages identified from SWIR spectroscopy include pyrophyllite+kaolinite, dickite+pyrophyllite, alunite+dickite, alunite+kaolinite and kaolinite+illite. Pyrophyllite, kaolinite and dickite are the dominant minerals of advanced argillic facies in the outcrop whereas alunite is observed only in the drill core section. Diaspore occurs locally in the advanced argillic alteration and generally associated with pyrophyllite.
Argillitic alteration is characterized by quartz, illite, paragonite, muscovite, kaolinite, smectite, illite-smectite and chlorite. The common assemblage is illite+montmorillonite. Propylitic alteration is composed of quartz, chlorite, epidote, illite, smectite, pyrite, hematite, goethite, carbonate, zeolite and magnetite. Phenocrysts and groundmass of host rock andesite lava and lapilli tuff are replaced by chlorite, sericite (illite), smectite, epidote, calcite, and albite. Chlorite and epidote tends to replace the mafic phenocrysts whereas as calcite and albite selectively replaces the plagioclase feldspar. Smectite generally replace the plagioclase and the groundmass is replaced by fine-grained chlorite, sericite, and epidote.

4 Conclusion

The application of reflectance spectroscopy enables to identify most hydrothermal alteration minerals related to the high-sulfidation mineralization in the Cijulang prospect. The minerals identified from SWIR spectroscopy help characterize the distinct alteration zonation which is significant in the determination of certain mineral deposit type.

Alteration system of the Cijulang prospect is similar to other systems that occur in well-documented high-sulfidation epithermal deposits of the world (e.g., Rodalquilar, Spain; Summitville, Colorado; Lepanto, Philippines). A characteristic alteration sequence and zonation of advanced argillic, argillic and propylitic alteration outward from the silicic core has resulted from the progressive cooling and neutralization of hot acidic magmatic fluid with the host rocks (Arribas, 1995). Formation of vuggy silica requires the fluids of at least pH \( \leq 2 \) and at a temperature of \( \sim 250^\circ C \) (Stoffregen, 1987) to leach all the minerals from the volcanic hosts except quartz. High-sulfidation epithermal system, which is characterized by advanced argillic alteration, has developed under relatively oxidized, acidic, \( SO_4 \)-dominated environment. Dominant pyrophyllite-kaolinite assemblage in the study area indicates that hy-
Figure 5: Hydrothermal alteration of Cijulang area, Garut Regency, West Java, Indonesia.
hydrothermal fluids responsible for the alteration.
were silica-saturated and the temperature of
formation may have been >260°C (White and
Hedenquist, 1990). The presence of pyrophyl-
lite is an indication of deep-seated formation of
the hydrothermal system.

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